

## Quiz on Learning Prerequisites – Solid State

Among the intermolecular forces, dipole-dipole force, London dispersion force, ionic interaction, and hydrogen bond, which is present in all substance?

- ☐ van der Waals force
- ☐ dipole-dipole force
- ☐ attractive force
- ☐ London dispersion force

Why do crystals diffract x-rays?

- ☐ X-rays have wavelengths of the order of the atomic spacing
- ☐ The used wavelength corresponds to the distance of the periodic lattice formed by the atoms in crystals
- ☐ The distance of the crystal planes is in the order of the wavelength.
- ☐ All of these.

The angle  $\theta$  of Bragg's law is

- ☐ the angle between the beam and the crystal
- ☐ the angle between the beam and the series of planes within the crystal which results in diffraction
- ☐ the angle between the crystal and the detector
- ☐ all of these

Strain is defined as the ratio of

- ☐ change in volume to original volume
- ☐ change in length to original length
- ☐ change in cross-sectional area to original cross-sectional area
- ☐ anyone of above
- ☐ none of these

Hook's law holds good up to

- ☐ yield point
- ☐ limit of proportionality
- ☐ breaking point
- ☐ elastic limit
- ☐ plastic limit

The ratio of direct stress to volumetric strain in case of a body subjected to three mutually perpendicular stresses of equal intensity, is equal to

- ☐ Young's modulus
- ☐ bulk modulus
- ☐ modulus of rigidity
- ☐ modulus of elasticity

Second order transitions are described by the Ehrenfest equations which describe a discontinuity for

- ☐ the entropy
- ☐ the heat capacity
- ☐ the isobaric thermal expansion coefficient
- ☐ the specific volume